

## BRIEF COMMUNICATIONS

## ORGANIC ACIDS OF MEDICINAL PLANTS.

3. *Matricaria recutita* and *M. matricaroides*

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Chamomile (*Matricaria recutita* L.) and pineapple weed [*M. matricaroides* (Less.) Porter] are known medicinal plants of the Asteraceae family. The first plant is certified. The second is used as a substitute [1]. The chemical composition of the biologically active substances in flowers of both species is well studied [2-4]. Only the phenolcarboxylic acids of the acid complex have been studied. Data on the di- and tricarboxylic acids (OA) of these species have not been reported.

Paper chromatography [5] found that the OA complexes of *M. recutita* and *M. matricaroides* flowers are identical and consist of five compounds. Separation of the OA fractions by preparative chromatography [6] isolated from *M. recutita* flowers five compounds that were identified as tartaric (Tar), citric (Cit), malic (Mal), malonic (Mln), and succinic (Suc) acids. Four compounds were isolated from *M. matricaroides*. Suc was not found. The chemical structures and purities were established by melting points of the compounds and their derivatives, melting-point depression of mixed samples, chromatographic mobilities, and  $^{13}\text{C}$  NMR spectroscopy.

The quantitative composition [7] showed that the total content of OA ( $X_{\Sigma}$ ) in both species was almost the same (7.15-7.37%) (Table 1). A large part of the OA were found in a bound form ( $X_b$ , 62-74%). Free-acid forms ( $X_f$ ) comprised less than one third of the total acid content. The average coefficients of the forms ( $k^f = X_f/X_b$ ) for *M. recutita* and *M. matricaroides* were 0.67 and 0.37%, respectively.

The specific profile of the OA (Table 2) showed that Tar was present in the studied species only as a bound form. Before ion exchange (IE), it could not be observed even with a significantly concentrated extract ( $k^f = 0$ ). Low coefficients were noted for Cit in *M. recutita* (0.18-0.20) and Mal in *M. matricaroides* (0.34-0.37). This indicates that both compounds in these samples are present primarily as bound forms.

Tar dominated the OA complexes in both instances. However, Suc dominated before IE in *M. recutita*; Cit, in *M. matricaroides*. The ratios of free acids Cit:Mal:Mln:Suc were 1.36:2.96:2.14:1.00 and 6.17:5.25:2.17:1.00 for *M. recutita* and *M. matricaroides*, respectively. The overall ratios Tar:Cit:Mal:Mln:Suc were 8.03:5.47:2.83:1.78:1.00 and 11.93:4.19:6.63:1.11:1.00, respectively. OA were isolated from *M. recutita* and *M. matricaroides* flowers for the first time.

TABLE 1. Group Composition of *M. recutita* and *M. matricaroides* Flower Acid Complex (min—max)/x

Species	$X_{\Sigma}$ , %	$X_f$ , %	$X_b$ , %	$k^f$
<i>M. recutita</i>	<u>7.16-7.30</u> 7.23	<u>2.21-2.83</u> 2.52	<u>4.47-5.03</u> 4.75	<u>0.63-0.74</u> 0.67
<i>M. matricaroides</i>	<u>7.15-7.37</u> 7.26	<u>1.91-2.00</u> 1.95	<u>5.15-5.44</u> 5.31	<u>0.36-0.38</u> 0.37

x is the average value;  $X_{\Sigma}$ , the total content of organic acids;  $X_f$ , the content of free-acid forms;  $X_b$ , the content of bound-acid forms;  $k^f$ , the form coefficients.

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TABLE 2. Specific Composition of *M. recutita* and *M. matricaroides* Flower Acid Complex (min—max)/x

Acid	<i>M. recutita</i>			<i>M. matricaroides</i>		
	before IE, %	after IE, %	k <sup>f</sup>	before IE, %	after IE, %	k <sup>f</sup>
Tartaric	-	<u>2.83-2.94</u> 2.89	0	-	<u>3.16-3.30</u> 3.22	0
Citric	<u>0.37-0.40</u> 0.38	<u>1.95-2.01</u> 1.97	<u>0.18-0.20</u> 0.18	<u>0.72-0.75</u> 0.74	<u>1.11-1.18</u> 1.13	<u>0.63-0.67</u> 0.64
Malic	<u>0.78-0.87</u> 0.83	<u>0.97-1.10</u> 1.02	<u>0.78-0.87</u> 0.81	<u>0.62-0.65</u> 0.63	<u>1.75-1.81</u> 1.79	<u>0.34-0.37</u> 0.35
Malonic	<u>0.60-0.62</u> 0.61	<u>0.35-0.45</u> 0.40	<u>0.58-0.75</u> 0.66	<u>0.24-0.27</u> 0.26	<u>0.25-0.35</u> 0.31	<u>0.74-0.96</u> 0.87
Succinic	<u>0.27-0.29</u> 0.26	<u>0.34-0.40</u> 0.36	<u>0.73-0.85</u> 0.79	<u>0.11-0.13</u> 0.12	<u>0.23-0.29</u> 0.27	<u>0.38-0.52</u> 0.45

x is the average value; before IE, content of OA before IE; after IE, content of OA after IE; k<sup>f</sup>, the form coefficients.

## REFERENCES

1. E. Arak, A. Raal, and J. Tammeorg, *Farmatsiya (Moscow, Russ. Fed.)*, **35**, No. 4, 19 (1986).
2. M. A. Prosovskii, K. S. Rybalko, V. I. Sheichenko, A. N. Shchavlinskii, and G. I. Oleshko, *Khim.-Farm. Zh.*, **19**, 981 (1985).
3. O. A. Konovalova and K. S. Rybalko, *Rastit. Resur.*, 116 (1982).
4. S. A. Chetvernaya, *Rastit. Resur.*, 373 (1986).
5. D. N. Olennikov, L. M. Tankhaeva, G. G. Nikolaeva, S. M. Nikolaev, A. V. Rokhin, and D. F. Kushnarev, *Khim. Prir. Soedin.*, 270 (2004).
6. D. N. Olennikov, T. M. Mikhailova, L. M. Tankhaeva, and A. B. Samuelsen, *Khim. Prir. Soedin.*, 378 (2005).
7. D. N. Olennikov, L. M. Tankhaeva, G. G. Nikolaeva, and A. A. Markaryan, *Rastit. Resur.*, 112 (2004).